



MICROCOPY RESOLUTION TEST CHART NATIONAL BUREAU OF STANDARDS-1963-A



AFGL-TR-86-0006

LASER SPECTROSCOPY OF OZONE

James L. Kinsey

Massachusetts Institute of Technology Department of Chemistry Cambridge, MA 02139

18 December 1985

SELECTE PRIS 1986

Final Report 1 July 1980 - 30 September 1985

APPROVED FOR PUBLIC RELEASE; DISTRIBUTION UNLIMITED

OTIC FILE COPY

AIR FORCE GEOPHYSICS LABORATORY
AIR FORCE SYSTEMS COMMAND
UNITED STATES AIR FORCE
HANSCOM AIR FORCE BASE, MASSACHUSETTS 01731

"This technical report has been reviewed and is approved for publication"

DANIEL H. KATAYAMA

Contract Manager

UV Surveillance & Remote Sensing Branch

LEON J. HEROUX

Chief, UV Surveillance & Remote

Sensing Branch

FOR THE COMMANDER

ROBERT A. SKRIVANEK

Director, Ionsopheric Physics Division

This report has been reviewed by the ESD Public Affairs Office (PA) and is releasable to the National Technical Information Service (NTIS).

Qualified requestors may obtain additional copies from the Defense Technical Information Center. All others should apply to the National Technical Information Service.

If your address has changed, or if you wish to be removed from the mailing list, or if the addressee is no longer employed by your organization, please notify AFGL/DAA, Hanscom AFB, MA 01731. This will assist us in maintaining a current mailing list.

Do not return copies of this report unless contractual obligations or notices on a specific document requires that it be returned.

ECURITY CLASSIFICATION OF THIS PAGE								
REPORT DOCUMENTATION PAGE								
1a REPORT SECURITY CLASSIFICATION UNCLASSIFIED		16. RESTRICTIVE MARKINGS						
24. SECURITY CLASSIFICATION AUTHORITY		3. DISTRIBUTION/AVAILABILITY OF REPORT Approved for public release; distribution unlimited						
2b. DECLASSIFICATION/DOWNGRADING SCHEDULE								
4. PERFORMING ORGANIZATION REPORT NUMBER(S)		5. MONITORING ORGANIZATION REPORT NUMBER(S)						
		AFGL-TR-86-0006						
Massachusetts Institute of Technology	Bb. OFFICE SYMBOL (If applicable)	7a. NAME OF MONITORING ORGANIZATION Air Force Geophysics Laboratory						
6c. ADDRESS (City, State and ZIP Code)	7b. ADDRESS (City, State and ZIP Code)							
Dept. of Chemistry Cambridge, MA. 02139		Hanscom AFB Massachusetts 01731-5000						
Se. NAME OF FUNDING/SPONSORING	86. OFFICE SYMBOL	9. PROCUREMENT I	NSTRUMENT ID	ENTIFICATION NU	MBER			
ORGANIZATION Air Force Geophysics Laboratory	(if applicable) LIU	F1 9628-80-к-00 88						
Sc. ADDRESS (City, State and ZIP Code)		10. SOURCE OF FUN	DING NOS.					
Hanscom AFB		PROGRAM ELEMENT NO.	PROJECT NO.	TASK NO.	WORK UNIT NO.			
Massachusetts 01731 11. TITLE (Include Security Classification)		61102F	2303	G1	AE			
Laser Spectroscopy of Ozone		<u> </u>						
12 PERSONAL AUTHOR(S) James L. Kinsey	- <u>-</u>							
13a. TYPE OF REPORT 13b. TIME COVERED FROM 1 Jul 800 30 Sep		14. DATE OF REPORT (Yr., Mo., Day) 1985 December 18 14		UNT				
16. SUPPLEMENTARY NOTATION 85								
	<u></u>							
17. COSATI CODES								
FIELD GROUP SUB. GR.		Ozone, spectroscopy, UV, fluorescence spectrum, Huggins bands, Hartley continuum.						
19. ABS RACT (Continue on reverse if necessary an	d identify by block number	•)						
Data have been obtained for the first time on the fluorescence spectrum of ozone in the Hartley continuum region. This type of data provides information for a new kind of spectroscopy with hitherto unprecedented capabilities for studying both ground state vibrational structure and excited state dissociation dynamics. From this data and theoretical models, qualitative generalizations are made about the potential energy surface of the upper electronic state of the Hartley transition. Experiments have also been conducted to obtain the high resolution spectrum of some of the lluggins bands for jet-cooled ozone. An analysis of the partially resolved rotational structure of these bands indicates that the symmetry of the Huggins bands is the same as that for the Hartley continuum.								
20. DISTRIBUTION/AVAILABILITY OF ABSTRATURE AND ASSTRATURE AND ASSESSED ASSESSED AND ASSESSED AND ASSESSED AND ASSESSED ASSESS		21. ABSTRACT SECURITY CLASSIFICATION UNCLASSIFIED						
22a. NAME OF RESPONSIBLE INDIVIDUAL				22c. OFFICE SYME	0			
Daniel H. Katayama		22b. TELEPHONE N (Include Area Ca (617)861-404	de)	LIU				

TABLE OF CONTENTS

Introduction	1
Personnel Associated With Work	6
Publications	6
Technical Presentations	7

Acces	ion For				
DTIC	ounced	9			
By					
Availability Codes					
Dist	Avail and/or Special				
A-1					



SUMMARY OF RESEARCH

This contract covered the period July 1, 1980 - September 30, 1985. The aim of the work was to improve understanding of the UV and visible spectrum of the atmospheric species 0_3 and to improve knowledge of its vibrational levels in the molecule's ground electronic state. Ozone photodissociates readily on absorption of light at any wavelength longer than $\sim\!850$ nm. Consequently, the absorption spectrum is entirely diffuse. There are three principal spectroscopic regions = The Chappuis band $(^1B_1+\tilde{\chi}^1A_1,\sim\!\lambda450-850$ nm), the Huggins band $(^1B_2+\tilde{\chi}^1A_1,\;\sim\!\lambda300-360$ nm) and the Hartley band $(^1B_2+\tilde{\chi}^1A_1,\;\sim\!\lambda200-300$ nm). The dominant and by far the most intense of these is the Hartley band, centered around $\sim\!\lambda255$ nm. Because of the rapid predissociation in this region, the band is almost entirely featureless, though there is some very faint structure near the band center. To the red, the much weaker (by $\sim\!10^{-3}$) Huggins system is also diffuse, but it exhibits definite vibrational structure.

The original idea of this work was to investigate the feeble structure in the Hartley band and the vibrational transitions in the Huggins band by looking at fluorescence excitation spectra of very cold ozone. The cooling (to ~10 K) was to be accomplished by adiabatic expansion of an ozone-helium mixture into a free-jet pulsed molecular beam. Our group already had some experience with 03 molecular beams and tuneable UV laser spectroscopy.

Introduction:

A summary of the research done for this contract is reported in this technical report. The detailed research results and the description of the experiments have been reported in the open literature and a listing of the journal articles is given in the Publication section of this report.

It is important to note that in the process of conducting laser experiments on ozone, fluorescence was unexpectedly observed when the laser was tuned to 266 mm which is in the peak absorption region of the Hartley continuum. This is surprising because it has always been assumed that the ozone is completely photodissociated in this wavelength region. Apparently, the laser light intensity is sufficiently strong that the small fraction of molecules that fluoresce is observable. This observation has led to a new kind of spectroscopy with unprecedented capabilities for studying excited state dissociation dynamics and ground state vibrational structure.

Shortly after the project started, these goals were expanded to include study of the resolved fluorescence of the photodissociating molecules. Inspired by theoretical ideas put forward by E. J. Heller and coworkers, we realized that such data could provide a new kind of spectroscopy with hitherto unprecedented capabilities for studying both ground-state vibrational structure and excited-state dissociation dynamics. The first measurements of this kind were made on ozone (in a static cell), and they revealed a sequence of vibrational levels in the ground electronic state reaching to within 500 cm⁻¹ of the molecule's dissociation limit. This was and remains a record for any polyatomic molecules.

The levels that could be assigned were all of the type $1_n 2_0 3_m$, i.e., no ground-state excited bending levels were observed. In the light of the Heller theory, this is a clear indication that dissociation on the upper optential energy surface proceeds with motion of the departing oxygen atom almost directly in the direction of the broken bond. Further qualitative generalizations were made about the shape of the excited-state potential in the Franck-Condon region. Specifically, it was concluded that the surface is saddle-shaped with respect to $q_1(\text{symmetric stretch})$ and $q_3(\text{antisymmetric stretch})$. This confirmed the idea that the weak structure in the Hartley band probably occurs because there are a few oscillations in the q_1 coordinate before the motion accelerates in the q_3 direction leading to dissociation. A more quantitative theoretical treatment has not yet been completed, but such a study is planned for the future. Later work also revealed a very small intensity into the q_1 level (~1% of that into q_2). The quantitative remeasurement of the whole q_3 emission spectrum, with special attention to

such minor features, will be an important step in the continuation of our studies on ozone.

A theoretical model for the stretching vibrations of ozone, based on a completely algebraic Hamiltonian, was completed in collaboration with Professor R. D. Levine. It was found that the algebraic approach could fit the observed levels competitively with conventional (Darling-Dennision) methods, yet retain a consistent anharmonic-oscillator perspective.

Darling-Dennison analysis mixes harmonic and anharmonic expressions.

Subsequent experimental work turned more towards the original idea, but focussed on the weak Huggins system instead of the Hartley band. A pulsed nozzle beam appropriate for these studies was developed and tested. At about this time, we also started considering whether some sort of optical multichannel analyzer could be found so that we could record an entire wavelength region simultaneously rather than by tedious scanning from line to line. Because of the extremely low light levels inherent in these experiments, it quickly became clear that the readout error would swamp the signal without at least two stages of amplification (image enhancement). Eventually, a unit that seemed suitable was identified and ordered (with funds from another source). Almost a year expired to delievery of the device, and it was found to have serious design problems as well as manufacturing defects. After a great deal of back and forth with the supplier, this instrument may now be just about ready to test for ozone work.

Meanwhile, the old technique was pursued in our investigation of the Huggins system. When excitation spectra were taken of jet-cooled ozone (the 501 and 600 vibrational bands), the widths of the features narrowed by a

factor of ~20 from the room-temperature spectra. This confirmed what we had suspected, that a large part of the room-temperature spectral width was due to rotational congestion. The new linewidth observation allowed a fresh estimate of the predissociative lifetime in the electronically excited state (~1 ps).

Further frequency narrowing of the excitation light actually permitted the partial resolution of rotational substructure in both vibrational bands, with individual rotational transitions estimated to be $\sim 1.4~{\rm cm}^{-1}$ wide. This increased the estimate of dissociative lifetime to ~4 ps. Because of incomplete thermal equilibration of the jet heam and the intrinsic width of individual rotational lines, these lines could not be fully resolved. Instead, the partially resolved structure was modelled with an asymmetric rotor band contour program. This showed quite clearly that both the $1_0^5 3_0^1$ and 1_0^6 vibrational transitions have the same rotational sturcture, which corresponds to an a-axis transition moment. If the equilibrium structure of the electronically excited state were of C_{2v} geometry, spin statistics would make this impossible: the existing rotational states would be a different set for the $1^{5}3^{1}$ level than for the 1^{6} level. Hence, we conclude that the electronically excited state has an equilibrium configuration only of Cs geometry, i.e., the two 0-0 bonds are of unequal length. In this case, for a high "inversion" barrier, the V3 quantum numbers would refer to single-well (left or right bond longer) levels. Each level, however, would be split into a closely spaced tunnelling doublet and one of the pair at any V3 would support the correct rotational states for a-type transitions. Since ab initio calculations indicate a C_S geometry for the $^{1}B_2$ excited state (and not for the $^{1}A_1$ state), we conclude that ${}^{1}B_{2}+\tilde{\chi}^{1}A_{1}$ is the correct electronic assignment for the

Huggins system. I.e., the Huggins band is actually just the wing of the Hartley hand, the former corresponding to transitions into bound regions of the potential energy surface and the latter to transitions into a Franck-Condon region with no excited-state quasibound levels. The details of the rotational band-contour analysis permitted estimates of $^{1}\text{B}_{2}$ rotational constants: A=2.1±.04 cm⁻¹, ^{1}B =0.45.01 cm⁻¹. Without data on isotopic variants these are not sufficient to establish a complete structure, but limits can be set on possible geometrics.

Personnel Associated With Work Performed Under

Contract F19628-80-K-0088

Principal Investigator

Current Position

James L. Kinsey

Professor of Chemistry, M.I.T.

Graduate Students

Dan Imre

Asst. Professor, U. of Washington

Amitabha Sinha

Postdoctoral Fellow, NOAA,

Boulder, CO.

Postdoctoral Associates

James H. Goble

Talandic Research, Pasadena, CA

Michael O. Hale

Postdoctoral Associate, M.I.T.

Publications:

CONTRACT ACCORDED TO THE CONTRACT OF THE CONTR

- 1. "Spectroscopic Characterization of Repulsive Potential Energy Surfaces: Fluorescence Spectrum of Ozone"
 - D. G. Imre, J. L. Kinsey, R. W. Field, D. H. Katayama
 - J. Phys. Chem. 86, 2564 (1982).
- "Chemical Dynamics Studied by Emission Spectroscopy of Dissociating Molecules"
 - D. Imre, J. L. Kinsey, A. Sinha and J. Krenos
 - J. Phys. Chem. 88, 3956 (1984).
- 3. "Excitation Spectroscopy of Jet-Cooled Ozone: The Huggins System" A. Sinha, D. Imre, J. H. Goble and J. L. Kinsey to be sumbitted to Journal of Chemical Physics.

Technical Presentations on Work Sponsored by Contract F19628-80-K-0088

Dam Imre:

ACS Annual Meeting 1985

Gordon Conference on Molecular Electronic Spectrosocopy 1985

University of Washington 1984 Princeton University 1984 University of Texas 1984 University of Pittsburgh 1984 University of Oregon 1984

Amitabha Sinha:

M.I.T. 1985

University of Colorado 1985

Air Force Geophysics Laboratory Ozone Meeting 1983

James L. Kinsey: University of Indiana - Gucker Memorial Lecture 1985

University of Illinois 1985

UCLA/USC/Caltech joint seminar 1985

University of Toronto 1985 University of Pennsylvania 1984

Harvard University 1984 Rutgers University 1984

XVI Informal Photochemistry Conference 1984

University of Rochester 1984

Dartmouth College 1984 Texas A&M University 1983 Rice University 1983 Tulane University 1983

University of Calfornia, Berkeley 1983

Stanford University 1983 University of Colorado 1983

Colby College 1983

DT/C 5-86

-